

UNITS AND MEASUREMENT

1.1 (a) Side of cube = 1 cm
 Volume of cube = (side)³
 $1 \times 1 \times 1 \text{ cm}^3$
 $= 1 \text{ cm}^3$
 $1 \times 1 \times 1 \text{ cm}^3 = \left(\frac{1}{100}\right)^3 \text{ m}^3$ $1 \text{ cm} = \frac{1}{100} \text{ m}$
 $= (10^{-2})^3$
 $= 10^{-6} \text{ m}^3$

(b) $r = 2.0 \text{ cm} = 2.0 \times 100 = 20 \text{ mm}$
 $h = 10.0 \text{ cm} = 10.0 \times 10 = 100 \text{ mm}$

Surface area of solid cylinder =
 $= 2\pi r^2 + 2\pi rh$
 $= 2\pi r(r+h)$

$= 2 \times \frac{22}{7} \times 20 (20 + 100) \text{ mm}^2$

$= \frac{4.8 \times 22 \times 100}{7}$

$= \frac{1056 \times 100}{7}$

$= 150.85 \times 100$

$= \cancel{15085} \text{ mm}^2 = 150.85 \times 10^2$

$= 1.5 \times 10^4 \text{ mm}^2$

(c) speed = 10 km/h

$= 10 \times \frac{5}{18} = 5 \text{ m/s}$

or
 $= 5 \text{ ms}^{-1}$

d). Relative density of lead = 11.3
 Relative density = $\frac{\text{density of substance}}{\text{density of water}}$
 $11.3 = \frac{\text{density of lead}}{1}$

$$\text{density of lead} = 11.3 \text{ g/cm}^3$$

In MKS system,

$$= 11.3 \times \frac{1}{1000} \times (100)^3$$

$$= 11.3 \times 10^{-3} \times 10^6$$

$$= 11.3 \times 10^3 \text{ kg m}^{-3}$$

$$= 1.13 \times 10^4 \text{ kg m}^{-3}$$

$$\left[\begin{array}{l} \therefore 1 \text{ cm} = \frac{1}{100} \text{ m} \\ \therefore 1 \text{ g} = \frac{1}{1000} \text{ kg} \end{array} \right]$$

1.2 Fill in the blanks:-

$$\begin{aligned} \text{(a)} \quad 1 \text{ kg m}^2 \text{ s}^{-2} &= 1 \times (1000 \text{ g}) \times (100 \text{ cm})^2 \times \text{s}^{-2} \\ &= 10^3 \text{ g} \times 10^4 \text{ cm}^2 \times \text{s}^{-2} \\ &= 10^7 \text{ g cm}^2 \text{ s}^{-2} \end{aligned}$$

$$\begin{aligned} \text{(b)} \quad 1 \text{ light year} &= 9.46 \times 10^{15} \text{ m} \\ \text{or } 1 \text{ m} &= \frac{1 \text{ light year}}{9.46 \times 10^{15}} \end{aligned}$$

$$1 \text{ m} = \frac{1}{9.46 \times 10^{15}}$$

$$= \frac{1 \times 10^{-15}}{9.46}$$

$$= \frac{10 \times 10^{-16}}{9.46}$$

$$= \frac{1000 \times 10^{-16}}{946}$$

$$\therefore 1 \text{ m} = 1.057 \times 10^{-16} \text{ ly}$$

$$\begin{aligned}
 (c) \quad 3.0 \text{ m/s}^2 &= 3 \text{ m/s}^2 \\
 &= 3 \times \frac{1}{1000} \text{ km} \times (60 \times 60)^2 \text{ hrs} \\
 &= \frac{3 \times 3600 \times 3600}{1000} \\
 &= 3 \times 360 \times 36 \\
 &= 38880 \text{ km H}^{-2} \\
 &\approx 3.9 \times 10^4 \text{ km H}^{-2}
 \end{aligned}$$

$$\begin{aligned}
 (d) \quad G &= 6.67 \times 10^{-11} \text{ Nm}^2 \text{ kg}^{-2} \\
 1 \text{ N} &= 10^5 \text{ dyne} \quad 1 \text{ m} = 100 \text{ cm} \quad 1 \text{ kg} = 1000 \text{ g} \\
 &= 6.67 \times 10^{-11} \times 10^5 \text{ dyne} \times (100 \text{ cm})^2 \times (1000 \text{ g})^{-2} \\
 &= 6.67 \times 10^{-11} \times 10^5 \times 10^4 \times 10^{-6} \text{ dyne cm}^2 \text{ g}^{-2} \\
 &= 6.67 \times 10^{-8} \text{ dyne cm}^2 \text{ g}^{-2}
 \end{aligned}$$

$$1.3. \quad 1 \text{ cal} = 4.2 \text{ J}$$

New system of unit =

Dimensional formula of energy $[ML^2T^{-2}]$

$$\therefore a=1, b=2, c=-2$$

Here,

$$[M_1] = 1 \text{ kg}, [L_1] = 1 \text{ m}, [T_1] = 1 \text{ sec}$$

$$[M_2] \propto m, [L_2] \propto m, [T_2] \propto s$$

$$n_1 = 4.2, n_2 = ?$$

we have formula,

$$\begin{aligned}
 n_2 &= n_1 \left[\frac{M_1}{M_2} \right]^a \left[\frac{L_1}{L_2} \right]^b \left[\frac{T_1}{T_2} \right]^c \\
 &= 4.2 \left[\frac{1 \text{ kg}}{\alpha \text{ kg}} \right]^1 \left[\frac{1 \text{ m}}{\beta \text{ m}} \right]^2 \left[\frac{1 \text{ s}}{\gamma \text{ s}} \right]^{-2} \\
 &= 4.2 \times \frac{1}{\alpha} \times \frac{1}{\beta^2} \times \frac{1}{\gamma^{-2}} \\
 &= 4.2 \alpha^{-1} \beta^{-2} \gamma^2 \text{ new units}
 \end{aligned}$$

1.5 Given that speed of light in vacuum is unity in new system of units

$$c = \frac{1 \text{ new unit of length}}{\text{sec}}$$

Time taken by light to reach earth = 8 min 20 sec
= 500 sec.

$$c = \frac{\text{distance}}{\text{time}}$$

$$\text{distance} = \text{speed} \times \text{time}$$

$$= 1 \times 500$$

$$= 500 \text{ units (new units)}$$

1.6 (c) The least count of an optical instrument is very low less in comparison to vernier calipers and screw gauge. Therefore optical instrument is most precise.

(c)

1.7 Linear magnification = $\frac{\text{Observed width}}{\text{Real width}}$

$$100 = \frac{3.5}{\text{Real width}}$$

$$\text{Real width} = \frac{3.5}{100}$$

$$= \frac{3.5}{100}$$

$$= 0.035$$

$$\text{Real width of hair} = \underline{\underline{0.035 \text{ mm}}}$$

1.8.

- (a) Wrap the thread on the metre scale with no gap between the threads. Measure the length l which contains n turns.

Diameter of thread,

$$d = \frac{l}{n}$$

- (b) we know that,

$$\text{least count} = \frac{\text{Pitch}}{\text{no. of divisions on the - circular scale.}}$$

If we increase no. of division, least count will decrease and accuracy increases but only to a certain limit as the resolution becomes low.

- (c) We know, to reduce random errors more or large no. of observations are taken. Therefore the set of 100 measurements is more reliable than the set of 5.

1.9 If $A = \text{area}$ and $L = \text{length}$ then,

$$\frac{A_1}{A_2} = \frac{L_2^2}{L_1^2}$$

or

$$\frac{L_2}{L_1} = \sqrt{\frac{A_2}{A_1}}$$

$$\text{here, } A_1 = 1.75 \text{ cm}^2 = 1.75 \times 10^{-4} \text{ m}^2$$

$$\text{and } A_2 = 1.55 \text{ m}^2$$

$$\frac{L_2}{L_1} = \sqrt{\frac{155}{175 \times 10^4}} = \sqrt{\frac{155}{175 \times 10^4}}$$

$$= \sqrt{\frac{155 \times 10^4}{175}}$$

$$= \sqrt{0.8857 \times 10^2}$$

$$= 0.9411 \times 10^2$$

$$= 94.11 \rightarrow 94.1$$

Hence the linear magnification is 94.1

1.10 (a) $0.007 \text{ m}^2 = 1$

(b) $2.64 \times 10^{24} \text{ kg} = 3$

(c) $0.2370 \text{ g cm}^{-3} = 4$

(d) $6.3205 = 4$

(e) $6.032 \text{ N m}^{-2} = 4$

(f) $0.0006032 \text{ m}^2 = 4$

1.11 $l = 4.234 \text{ m}$ $b = 1.005 \text{ m}$

$t = 2.01 \text{ cm} = 0.0201 \text{ m}$

Area of metal sheet

$$A = 2(lb + bh + hl)$$

$$= 2 [(4.234 \times 1.005) + (1.005 \times 0.0201) + (0.0201 \times 4.234)]$$

$$= 2 [4.25517 + 0.0202005 + 0.0851034]$$

$$= 2 \times 4.3604739$$

$$= 8.720947 \text{ m}^2$$

or 8.72 m^2

Now,

$$\text{Volume} = l \times b \times t$$

$$\begin{aligned}
 &= 4.234 \times 1.005 \times 0.0201 \\
 &= 0.0855289 \\
 &= 0.0855 \text{ m}^3
 \end{aligned}$$

1.12

Mass of grocery box = 2.300 kg
 Mass of gold piece 1 = 20.15 g
 Mass of gold piece 2 = 20.17 g
 $= 0.02017 \text{ kg}$

(a). Total mass of the box with gold pieces

$$\begin{aligned}
 &= 2.3 + 0.02015 + 0.02017 \\
 &= 2.34032 \\
 \text{or } &= 2.3 \text{ kg}
 \end{aligned}$$

(b) Difference in masses of gold pieces

$$\begin{aligned}
 &= 20.17 - 20.15 \\
 &= 0.02 \text{ g}
 \end{aligned}$$

1.13

The boy writes the relation as,

$$m = \frac{m_0}{(1-u^2)^{1/2}}, \quad \begin{array}{l} m \rightarrow \text{moving mass} \\ m_0 \rightarrow \text{rest mass} \\ u \rightarrow \text{velocity} \end{array}$$

check the correctness of formula,

$$[M] = \frac{[M]}{[LT^{-1}]}$$

$$\text{LHS} \neq \text{RHS}$$

Hence, the formula is incorrect

for correction we need to put c^2 in division of v^2 , then the denominator will become dimensionless and both sides will be of same dimension. i.e.,

$$m = \frac{m_0}{\left(1 - \frac{v^2}{c^2}\right)^{1/2}}$$

1.14 The size of hydrogen atom = $0.5 \text{ \AA}$
i.e. $r = 0.5 \text{ \AA}$
 $= 0.5 \times 10^{-10} \text{ m}$

Volume of hydrogen atom,

$$\begin{aligned} \frac{4}{3} \pi r^3 &= \frac{4}{3} \times 3.14 \times (0.5 \times 10^{-10})^3 \\ &= \frac{4}{3} \times 3.14 \times 0.125 \times 10^{-30} \\ &= \frac{3.14 \times 0.5}{3} \times 10^{-30} \\ &= 0.523 \times 10^{-30} \text{ m}^3 \end{aligned}$$

Now volume of one mole of hydrogen atom,

$$\begin{aligned} &= 6.023 \times 10^{23} \times 0.523 \times 10^{-30} \\ &= 3.15 \times 10^{-7} \text{ m}^3. \\ &\approx 3 \times 10^{-7} \text{ m}^3 \end{aligned}$$

1.15 radius of the hydrogen molecule =

$$r = 1 \text{ \AA} = 1 \times 10^{-10} \text{ m}$$

Volume of hydrogen molecule,

$$V = \frac{4}{3} \pi r^3$$

$$= \frac{4 \times 3.14 \times (1 \times 10^{-10})^3}{3}$$

$$= \frac{12.56 \times 10^{-30}}{3}$$

$$= 4.189 \times 10^{-30} \text{ m}^3$$

Now,

We know, $N_A = 6.023 \times 10^{23}$ atom

$$\text{Atomic volume} = 4.189 \times 10^{-30} \times \frac{6.023}{10^{23}}$$

$$= 25.23 \times 10^{-7} \text{ m}^3$$

Molar volume = 22.4 l

$$= 22.4 \times 10^{-3} \text{ m}^3$$

$$\left[1 \text{ l} = \frac{1}{1000} \text{ m}^3 \right]$$

Ratio,

$$\frac{\text{Molar volume}}{\text{Atomic volume}} = \frac{22.4 \times 10^{-3}}{25.23 \times 10^{-7}}$$

$$= 0.89 \times 10^4 = 10^4$$

The ratio is very large because the size of a gas molecule is very small in comparison with the volume of the gas.

1.16

The direction of line of sight changes very rapidly for nearby objects like trees, houses etc. therefore, they appear moving fast. However, line of sight does not change much for distant object like hills, stars, moon etc. therefore they appear stationary.

1.17

Given,

$$\text{Mass of sun} = 2 \times 10^{30} \text{ kg}$$

$$\text{Radius of sun} = 7 \times 10^8 \text{ m}$$

$$\text{Density} = \frac{\text{Mass}}{\text{Volume}}$$

$$= \frac{2 \times 10^{30}}{\frac{4}{3} \pi \times (7 \times 10^8)^3}$$

$$= \frac{3 \times 10^{30}}{2 \times \pi \times 7 \times 7 \times 7 \times 10^{24}}$$

$$= \frac{3 \times 10^{30}}{2 \times 22 \times 7 \times 49 \times 10^{24}}$$

$$= \frac{3 \times 10^6}{2156}$$

$$= \frac{3 \times 10^3}{2.156}$$

$$= 1.38 \times 10^3$$

$$\text{density} = 1.4 \times 10^3 \text{ kg/m}^3$$